

The **dry adiabatic lapse rate** (DALR) describes the rate at which the temperature of a parcel of unsaturated (dry) air decreases as it rises in the atmosphere under adiabatic conditions—no heat exchange with the surroundings. It is conventionally given as approximately **9.8 °C/km** (or more precisely ~9.76–9.8 K/km, with the lapse rate defined as the positive value $-dT/dz$).

This value is **not a coincidence** with the acceleration due to gravity $g \approx 9.8 \text{ m/s}^2$; it arises directly from the physics of the process. The DALR equals g / c_p , where c_p is the specific heat capacity of dry air at constant pressure (approximately 1004–1005 J/(kg·K) for Earth's atmosphere).

Derivation (Step by Step)

Consider a parcel of dry air rising in the atmosphere. Two key equations combine here:

1. **Hydrostatic equilibrium** (balance of gravity and pressure gradient in the atmosphere):

$$dP = -\rho g dz$$

(Pressure decreases with height dz due to the weight of the air above; ρ is density, g is gravitational acceleration.)

2. **First law of thermodynamics for an adiabatic process** ($dQ = 0$, no heat added or removed):

The parcel expands as pressure drops, doing work and cooling internally. For dry air treated as an ideal gas, this leads to:

$$c_p dT = \alpha dP$$

(where $\alpha = 1/\rho$ is the specific volume). More precisely, the enthalpy form simplifies to $c_p dT = v dP$ (or equivalent after using the ideal gas law).

Combining them (substituting dP from hydrostatic balance into the thermodynamic equation and rearranging):

$$dT/dz = -g / c_p$$

Thus, the lapse rate $\Gamma_d = -dT/dz = g / c_p$.

For Earth:

- $g \approx 9.81 \text{ m/s}^2$

- $c_p \approx 1004 \text{ J}/(\text{kg}\cdot\text{K}) = 1004 \text{ m}^2/(\text{s}^2\cdot\text{K})$ (since $1 \text{ J} = 1 \text{ kg}\cdot\text{m}^2/\text{s}^2$, so units simplify cleanly)

Dividing:

$g / c_p \approx 9.81 / 1004 \approx 0.00977 \text{ K/m} = \mathbf{9.77 \text{ K/km}}$ (often rounded to $9.8 \text{ }^\circ\text{C/km}$, as $^\circ\text{C}$ and K differences are identical).

The numerical match with $g = 9.8$ occurs because c_p for dry air is very close to **1000 $\text{J}/(\text{kg}\cdot\text{K})$** (a convenient round number in approximations), and converting from per meter to per kilometer introduces a factor of 1000 that cancels nicely with the $\sim 9.8 / \sim 1000$ ratio.

The units work out exactly to K/km (or $^\circ\text{C/km}$) because:

- g has units m/s^2
- c_p has units $\text{m}^2/(\text{s}^2\cdot\text{K})$
- g / c_p has units $\text{K/m} \rightarrow$ multiply by 1000 for K/km

This relation holds generally for any planetary atmosphere (dry case); the value changes with that planet's g and the c_p of its atmosphere (e.g., it is higher on Venus due to its g and different c_p).

Physical Intuition

- As the air parcel rises, it encounters lower pressure and expands.
- Expansion requires the parcel to do work on its surroundings (pushing neighboring air aside).
- With no heat input (adiabatic), this work comes from the parcel's internal energy $\rightarrow$ temperature drops.
- The rate depends on how much gravitational potential energy change (tied to g via hydrostatic balance) must be balanced by the cooling needed to maintain the expansion work (tied to c_p , which measures how much temperature change is needed to change the air's energy).

The **moist adiabatic lapse rate** is smaller (typically $4\text{--}9 \text{ }^\circ\text{C/km}$) because condensation releases latent heat, partially offsetting the cooling.

This 9.8 value is a fundamental consequence of thermodynamics and hydrostatics in a gravitational field—not a coincidence, but a direct outcome where Earth's g and air's c_p

gravitational field. Not a coincidence, but a direct outcome where Earth's g and u_p happen to produce a clean number.